The Metropolitan

Water Reclamation District of Greater Chicago

## WELCOME TO THE JULY EDITION OF THE 2010 M&R SEMINAR SERIES

## BEFORE WE BEGIN

- SILENCE CELL PHONES & PAGERS
- QUESTION AND ANSWER SESSION WILL FOLLOW PRESENTATION
- SEMINAR SLIDES WILL BE POSTED ON MWRD WEBSITE AT (www. MWRD.org)
- Home Page → (Public Interest) → more public interes
   → M&R Seminar Series → 2010 Seminar Series

Greenhouse Gas (GHG) Accounting for the Metropolitan Water Reclamation District of Greater Chicago's Biosolids Management Program

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Sources: Center for climatic research, Institute for environmental studies, university of Wisconsin at Madison; Okanagan university college in Canada, Department of geography; World Watch, November-December 1998; Climate change 1995, The science of climate change, contribution of working group 1 to the second assessment report of the intergovernmental panel on climate change, UNEP and WMO, Cambridge press university, 1996.

#### Short term carbon cycle- soils and plants



Energy from the sun is used to 'fix' atmospheric CO<sub>2</sub> via photosynthesis



The fixed  $CO_2$ , in the form of plant matter is used as food by a wide range of animals including microorganisms

A portion of the carbon remains fixed (soil organic matter animal biomass) and the remainder decomposes aerobically and returns to the atmosphere as  $CO_2$ 



## Magnitude of different cycles



Focus







#### However, Land disturbance and Climate Change





### Wastewater treatment

- A much more engineered system than the bear in the woods
- But basically part of the short term carbon cycle



# The human version of the short term carbon cycle





## So for example

- Biological treatment is a natural process that is greatly expedited by pumping air
- Energy required for the air is a GHG debit



## Anaerobic digestion-Can be a significant source of credits

- 2260 kWh dry Mg biosolids
- Co-digestion will significantly increase this



## Biosolids

Can impact the carbon cycle:

- By making or using energy
- By replacing products that require energy to produce
- By replacing products that emit GHGs
- By emitting gasses other than CO2
- By sequestering carbon

#### Combustion: potential source of credits (Metcalf and Eddy, 100% efficiency)



<sup>%</sup> Solids

#### Gasses other than $CO_2$ can have a huge impact $CH_4$ -21-23 x $CO_2$ N<sub>2</sub>O 296 x $CO_2$



Freeboard temperature [K]

## So for one fluidized bed mono-combustion facility



А

#### Biosolids GHG spreadsheet calculator tool

- Funded by the CCME
- Andrew Carpenter, Northern Tilth
- Ned Beecher, NEBRA





## Spreadsheet

- Storage
- Conditioning/ thickening
- Aerobic/anaerobic digestion
- Landfill disposal
- De-watering
- Thermal drying
- Alkaline stabilization
- Combustion
- Land application
- Transportation

· · · · ·		
Unit Processes & Inputs	Inputs & Daily Emissions	Default Input (Optional)
Solids Input (to incinerator)		
Quantity (Mg/day-wet)	100	
Solids content (%)	25.0%	
Quantity (Mgiday-dry)	25.0	
Table digested prior to increation?	10%	4.0%
Total shoeshorus (%-dry weight)	15%	1.5%
TVS(%-dry weight)	70.0%	70.0%
Type of incinerator	Fluidized Bed	Fluidized Bed
Recovered energy to electricity (%)	0%	
Recovered energy as heat (%)	75%	
Disposition of ash - Is it used to replace phosphorus fertilizer or in cement or brick?	none	0000
is a yrea-based selective noncatalytic reduction emissions system being used?	no	no
Average high (freeboard) temperature of combustion (*C)	850	850
Energy Balance		
CIOPS Trigger and the incinerator-evaporating water (BtL/day)		319,688,238
Energy potential of sludge (Btulday)		544,994,775
E-state-		
Puel Use		0.054
Survided cos use from seconosed energy (m <sup>2</sup> /day)		9,041
Not natural nas used (m <sup>2</sup> May).	-198	-198
CO <sub>2</sub> emissions from natural cas used (Molday)	-0.37	-120
Electricity Use		
Electricity requirements of incinerator (kWh/day)		5,000
Electricity generated (kWh/day)		0
Net Electricity used (kWh/day)	5,000	5,000
CO <sub>2</sub> emissions from electricity used (Mg/day)	0.91	
ion		
Netweenissions		
CO2 emissions equivalents from released methane (Mg/day)	0.03	
Marcon Andre Frederiken		
Nitrous Oxide Emissions	0.064	0.054
N <sub>2</sub> O emission anti-stream for SNC9 based on use (Moldar)	0.000	0.054
N-O emission adjustment for moisture content of shuites (Making)	-0.032	
CO., emissions equivalents from released N.O (Molday)	9.92	
Cement Replacement Value		
CO <sub>2</sub> replacement value from cement manufacture (Mg CO <sub>2</sub> /day)	0.00	
Fertilizer Off-set Credits		
From phosphorus applied to soil (Mg CO <sub>2</sub> /day)	0.00	
Biomass Combustion	25.82	
<ul> <li>cmissions equivalents from outning studge (Mg/day)</li> </ul>	35.93	
CO <sub>2</sub> equivalents (Melveer)	3 825	
Correctivents (mgrycar)	3,454	
Scope 1	331	
Scope 2 Research 1 & 2	3,825	
beoper to a		

#### First - not discussed Drying- Centrifuge

- Centrifuge
- Higher % solids
- Much higher energy use
- 0.04-0.2 kWh/m<sup>3</sup> wet
- Higher emissions from energy use



- Belt filter press
- Lower % solids
- Much lower energy use
- 0.004-0.01 kWh/m<sup>3</sup> wet



## Polymer

- 23 Mg CO<sub>2</sub> per Mg polymer
- Average use of 5 kg per dry Mg biosolids
- 110 kg CO<sub>2</sub> per dry Mg biosolids to aid in dewatering



Applied the tool to calculate GHG balance for **MWRD's** biosolids program





CONSTRUCTING PARKS, GOLF COURSES, AND ATHLETIC FIELDS USING BIOSOLIDS



## MWRD Program

- Focus on biosolids
- Data from 2 years
   2001,2008
- Debits
- Credits
- Unknowns



## Where biosolids went

<b>Biosolids Use</b>	2001	2008	
<b>Class B Dewatered cake</b>			
Farmland Fertilizer	128,100	97,100	
Landfill Daily Cover	35,700	26,500	
<b>Class A Air-Dried</b>			
Urban Reclamation	3,100	20,000	
Mineland Reclamation	22,000	0	
Landfill Final Cover	1,500	46,500	
Landfill Co-Disposal	12,700	2,000	

Total	203,100	192,100

# Chemical and physical characteristics of biosolids

	<b>Air-Dried</b>		Dewater	<b>Dewatered</b> Cake	
	2001	2008	2001	2008	
Total Solids (%)	70.2	71.3	25.9	25.3	
<b>Total Volatile Solids (%)</b>	35.4	38.1	54.2	52.5	
Total Kjeldahl N (%)	1.87	2.22	4.29	4.18	
NH4 <sup>+</sup> - N (%)	0.40	0.20	0.43	0.54	
NO <sub>3</sub> <sup>-</sup> - N (%)	0.044	0.031	ND	0.007	
Total P (%)	2.20	2.13	2.52	1.94	
Bulk Density (g cm <sup>-3</sup> )	0.68	ND	0.52	ND	

## Fulton County biosolids land reclamation







## Strip-mined land



#### 1973

# From liquid to dewatered and air-dried biosolids (1972-2004)





## Biosolids-amended land (2009)





## Biosolids build up Soil Org. C





## Soil organic C response to biosolids



Cumulative biosolids application rate (dry Mg ha<sup>-1</sup>)

Tian et al. 2009. JEQ 38: 61-74.

## **Biosolids-amended soil**



## H1: direct addition to stable SOC

Biosolids high in Alkyl C, mainly microbial transformation of O-alkyl to Alkyl during the aeration.

- O-alkyl C: easily decomposable
- Alkyl C: resistant to decomposition

#### Carbon functional groups of humic acids in FC soils detected by Nuclear Magnetic Resonance (NMR)



H2: Biosolids Fe/Al reduces the oxidation of crop residue-derived SOC

- Fe(OH)<sub>n</sub><sup>+</sup>
- R-COO

Fe-soil organic matter complexes
 Co-precipitation (Schwertmann et al. 2005)
 Adsorption (Kaiser et al. 2007)

#### Fe (OH)<sub>n</sub> coating decreases the access of biota to SOC



## Organically bound Fe in FC soils



#### Track the Biosolids C Remaining

- Exponential decay model:  $y = a_1 e^{-k^{1t}} + a_2 e^{-k^{2t}} + a_3 e^{-k^{3t}}$
- Fractions: labile (k<sub>1</sub>), slow (k<sub>2</sub>), recalcitrant (k<sub>3</sub>)



Remaining of initial mass (%)

 k adjusted to Fulton county climate conditions using ratio derived from Gilmour and Gilmour. 1980. JEQ 9: 194-199.

#### Biosolids C remaining equation

$$y = [0.35 - (55 - Vs)/55]C_{BS}e^{-0.0222t} + [0.52 + 0.8*(55-Vs)/55]C_{BS}e^{-0.000381t} + [0.13 + 0.2*(55 - Vs)/55]C_{BS}e^{-0.0000448t}$$

y = remaining biosolids C in soil (Mg C ha<sup>-1</sup>) Vs = volatile solids in biosolids (%) C<sub>BS</sub> = C input from applied biosolids (Mg C ha<sup>-1</sup>) t = time (day)

## Mean residence time of biosolids C



## Soil C sequestration rate at Fulton county (Mg C ha<sup>-1</sup>, 22.4 DT biosolids)



#### We used recalcitrant C fraction as undecomposable biosolids C

- Liquid (vs = 55%)  $f_u = 0.13$
- Dewatered cake (vs = 52.5%), f<sub>u</sub> = 0.14
- Air-dried (vs = 35.8%)

 $f_{u} = 0.2$ 

## Landfill- Class B Disposal and daily cover

Credits

Carbon storage

#### Debits

- Transport
- Methane emissions
- Nitrous oxide emissions



## Landfill emissions



## Landfill emissions



## Landfill- Final cover

Class A dry material

 Used at final surface
 Used as a soil material



## Different picture



Mg Dry Biosolids

## 2001 versus 2008

- CH<sub>4</sub> and N<sub>2</sub>O debits are highly significant
- Transport minimal
- Credit potential for final cover



## Agricultural use

- Benefits
  - Fertilizer
     offsets
  - Soil carbon
     storage
- Debits

   Transport
   N<sub>2</sub>O?



## Agriculture- Class B



## Class A urban versus Class B ag







6 of one....



#### End use options- GHG emissions



## Uncertainties

- CH<sub>4</sub> likely to be minimal in an aerobic soil environment
- N<sub>2</sub>O currently treated to be equivalent to fertilizer emissions
  - Increase in  $N_2O$  could result in a significant debit

## Preliminary studies

- Biosolids added to turf in the greenhouse
- Surface application of Class A cake
- Sandy soil
- Clay soil

## Higher $N_2O$ from biosolids



Day

## Higher $N_2O$ from biosolids



## Drying beds

- 50% of the total N is lost
- 20 kg per dry Mg
- If 10% is lost as N<sub>2</sub>O
- Debit of 600 kg per dry Mg biosolids

## Conclusions

- Biosolids can be a significant source of GHG credits
- Transport is of minimal importance
- Quantifying C sequestration is important
- Understanding and minimizing fugitive gas emissions is critical