Fundamental and practical studies on Enhanced Biological Phosphorus Removal (EBPR)

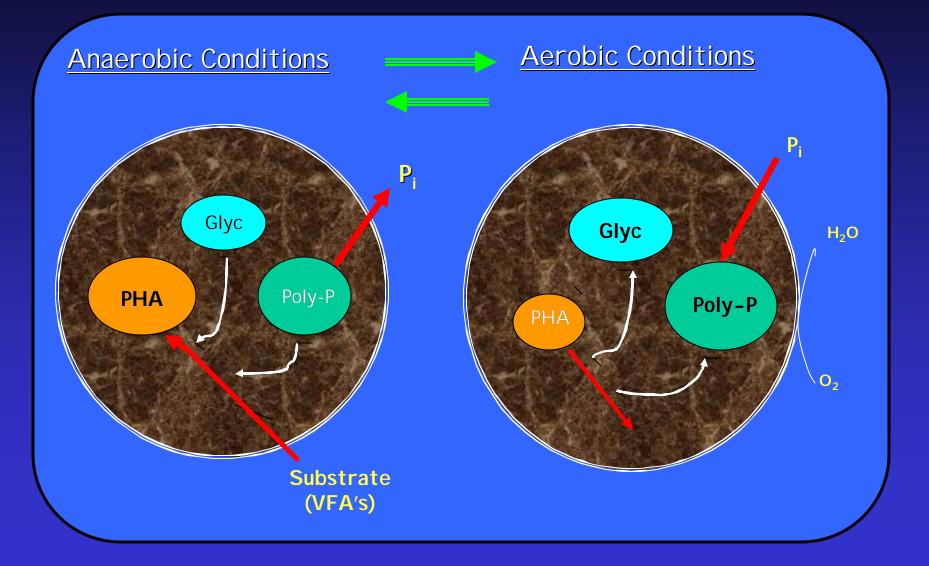
I dentifying polyphosphate accumulating organisms and achieving very low effluent phosphorus concentrations

Daniel R. Noguera

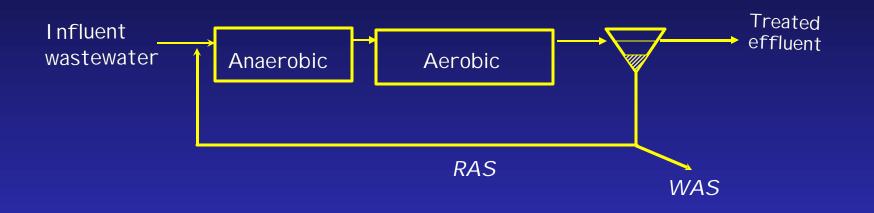
Department of Civil and Environmental Engineering University of Wisconsin – Madison

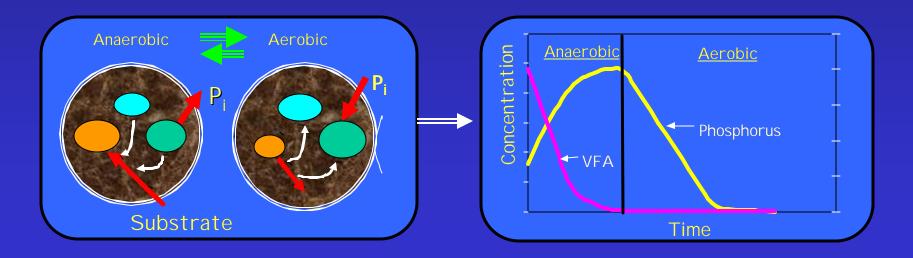
Chicago, March 30, 2007

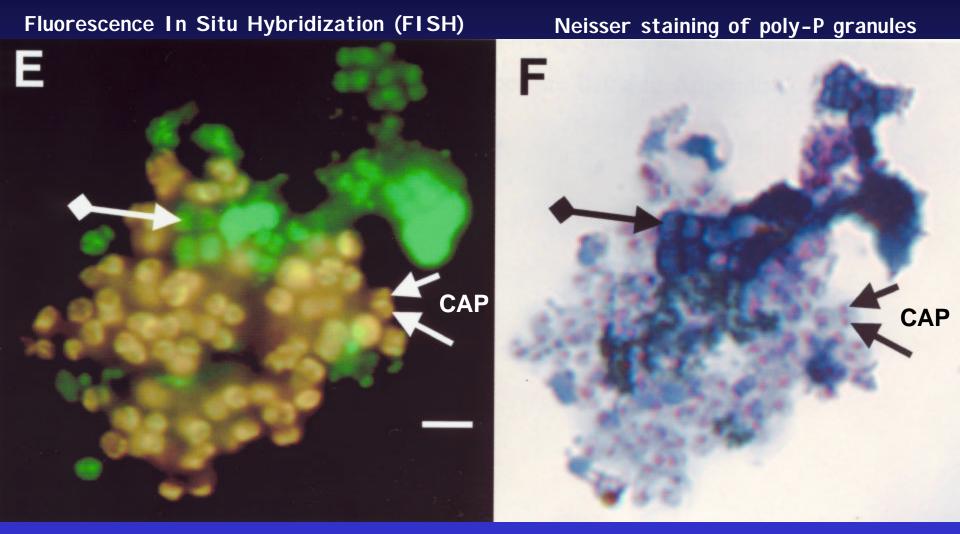
Enhanced biological phosphate removal (EBPR)



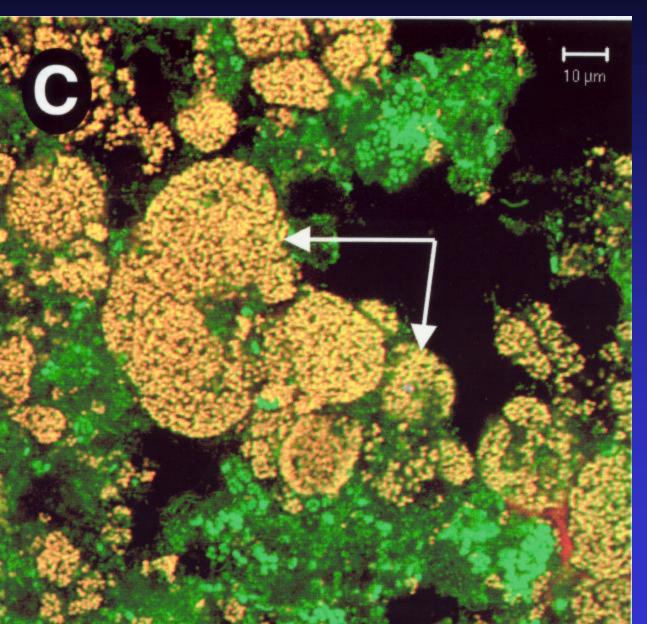
Enhanced Biological Phosphate Removal (EBPR)







Crocetti et al. 2000 - AEM

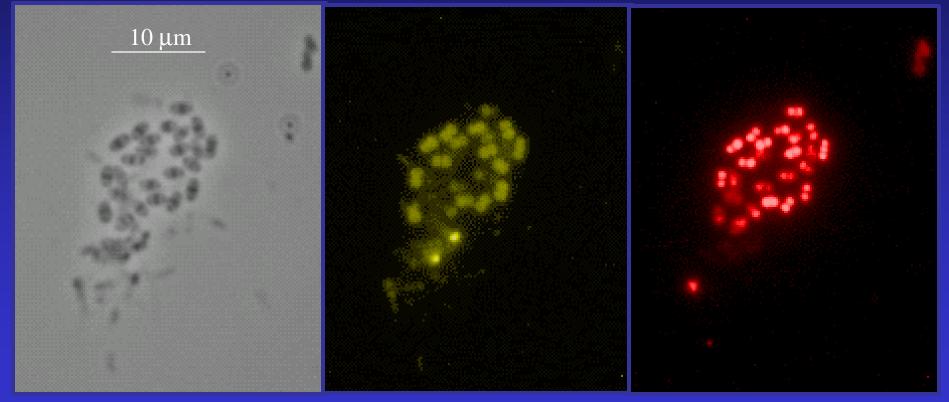


Crocetti et al. 2000 - AEM

Phase contrast

Polyphosphate (DAPI staining)

FISH



Zilles et al. 2002 - WST



<u>SBR</u>

6 hour cycles 2h anaerobic 3h aerobic 1h settling/decant HRT = 18 hours SRT = 10 days pH = 7.2Fed: Acetate = 460 mg/L P = 35 mgP/LAllylthiourea

Enrichment cultures – UW ~ 80% CAP

CAP in full-scale EBPR plants

What is the concentration of CAP in full-scale plants?

What is its contribution to phosphorus removal?

Is CAP contribution the same in all EBPR treatment plants?

Full-scale quantification

CAP in full-scale EBPR plants

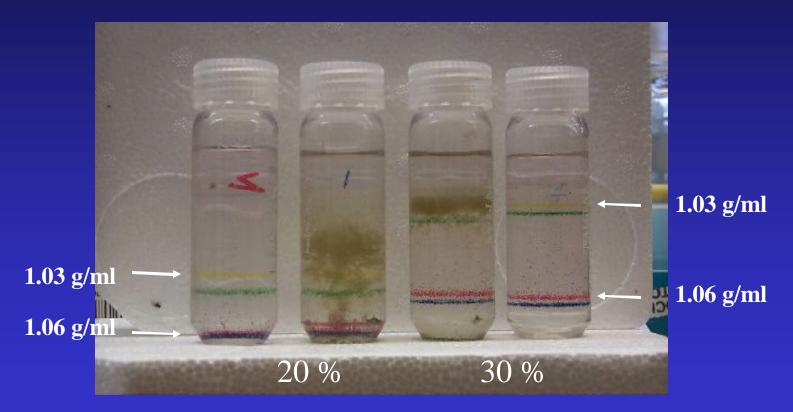
FISH alone does not tell you whether the organism is participating in EBPR

Combining FISH with Neisser staining or DAPI staining is technically difficult

Can PAOs be physically separated from other microorganisms in activated sludge floc? Density centrifugation Flow cytometry

Separation of PAOs by density centrifugation

PAO are heavier than other organisms in activated sludge



Percoll gradients

Hung et al. 2002 - WER

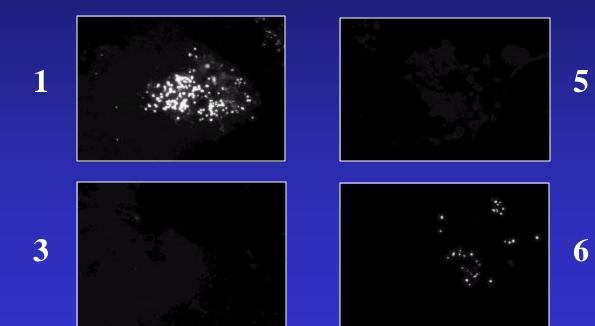
Separation of PAOs by density centrifugation

PAO are heavier than other organisms in activated sludge

After centrifugation

Microscopic visualization of DAPI stained cells

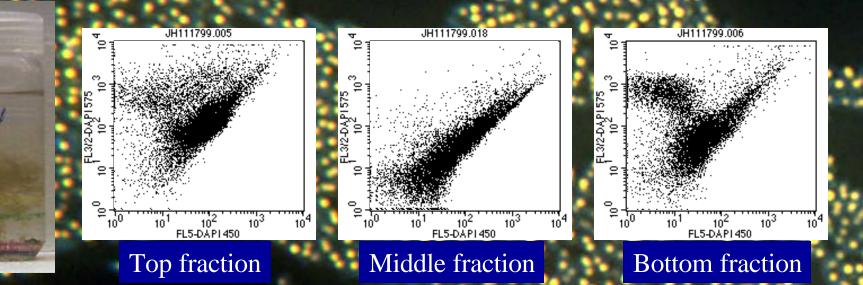




Hung et al. 2002 - WER

Separation of PAOs by cell sorting after DAPI staining (and after density centrifugation)

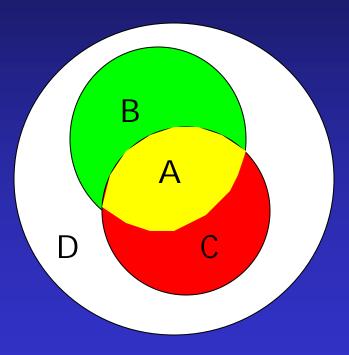
With DAPI, PAOs have yellow fluorescence, while all other cells have blue fluorescence



Hung et al. 2002 - WER

Quantification Methodology

How relevant is CAP in full-scale WWTPs?



A = FISH positive and PolyP positive

- **B** = **FISH** positive and PolyP negative
- **C** = **FISH** negative and **PolyP** positive
- **D** = **FISH** negative and PolyP negative

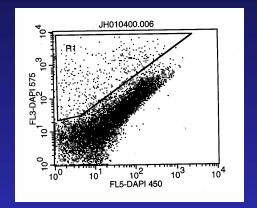
A + B + C + D = 100%

PolyP + FI SH +

Quantification Methodology

How relevant is CAP in full-scale WWTPs?

In PolyP-rich fraction:



n = concentration factor for PolyP cells m = concentration factor for non-PolyP cells nA + mB + nC + mD = 100%

Quantification Methodology

How relevant is CAP in full-scale WWTPs?

Activated Sludge: A + A +

A + B + C + D = 100% A + B = FISH positiveA + C = PolyP positive

PolyP-rich fraction: nA + mB + nC + mD = 100% nA + mB = FISH positive nA+ nC = PolyP positive

Nine Springs Wastewater Treatment Plant - Madison, WI (42 MGD, UCT process w/o nitrate recycle)

B	Quantification Results Nine Springs WWTP (UCT process)				
DC	FISH positive		PolyP ositive		
Activated Sludge:	18 %	2	28 %		
PolyP-rich fraction:	41 %	5	8 %		
A = 0.205 B = -0.025	C = 0.075 D = 0.745	n = 2.07 m = 0.58			

• $A/(A+C) = 0.73 \rightarrow 73\%$ of PolyP are CAP

• $A/(A+B) = 1.1 \rightarrow 100\%$ of CAP have PolyP

CAP in full-scale EBPR plants

What is the concentration of CAP in full-scale plants?

What is its contribution to phosphorus removal?

Is CAP contribution the same in all EBPR treatment plants?

Full-scale quantification

Dane-I owa Wastewater Treatment Plant - Mazomanie, WI (0.5 MGD, Orbal process)

1220

. 100

.

В	_	Quantification Results				
A	Da	Dane-I owa WWTP (Orbal process)				
D C		FISH		PolyP		
		positive	р	ositive		
Activated Sludge:		13 %	2	2 %		
PolyP-rich fraction:		18 %	ļ	52 %		
A = 0		C 0162	n = 2.36			
A = 0 $B = 0.$		C = 0.163 D = 0.707	m = 2.30 m = 0.61			

• $A/(A+C) = 0.26 \rightarrow 26\%$ of PolyP are CAP

• $A/(A+B) = 0.44 \rightarrow 44\%$ of CAP have PolyP

Searching for other PAOs

Lab-scale simulation of aerated-anoxic EBPR processes



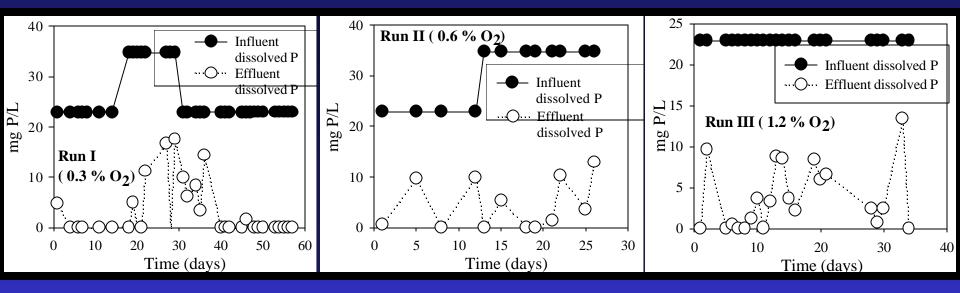
<u>SBR</u>

6 hour cycles 2h anaerobic 3h aerobic 1h settling/decant HRT = 18 hours SRT = 10 days pH = 7.2Fed: Acetate = 460 mg/L P = 35 mgP/LAllylthiourea

Anaerobic stage converted to aeratedanaerobic by adding $0.3\% O_2$ to gas supply

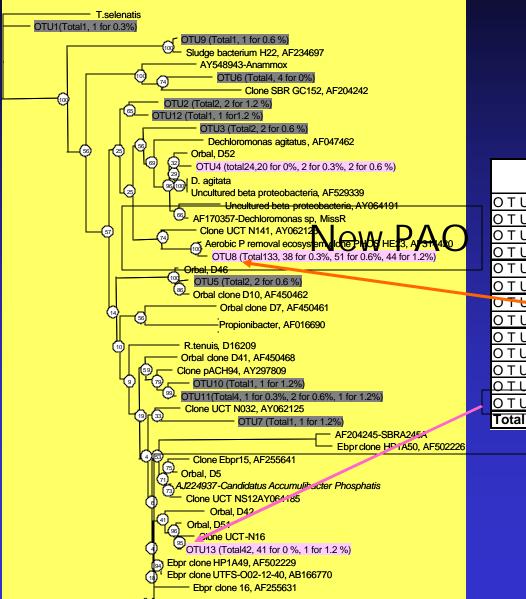
Searching for other PAOs

Lab-scale simulation of aerated-anoxic EBPR processes



Searching for other PAOs 16S rRNA cloning of aerated-anoxic EBPR sludge ~ 1,500 bases **16S rRNA FISH probe Cloned fragment** labeled dNTP's sequencing reaction 16S rDNA 110 AATGACCGGTAACCCGTAACTAA ~600 nucleotides long sequence

Searching for other PAOs



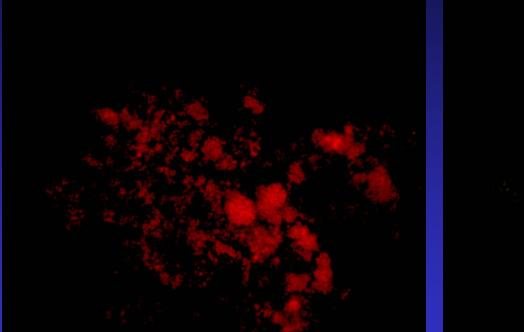
Phylogenetic analysis

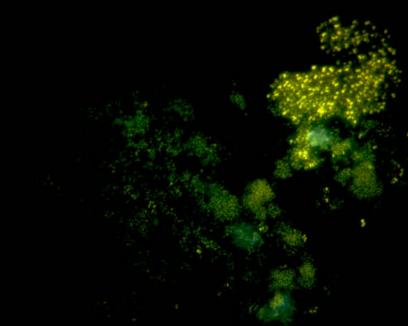
	Total Number of clones covered by this O T U							
	0 % DO	0.3 % DO	0.6 % DO	1.2 % DO	Total			
OTU1	0	1	0	0	1			
OTU2	0	0	0	2	2			
ОТИЗ	0	0	2	0	2			
OTU4	20	2	2	0	24			
O T U 5	0	0	2	0	2			
OTU6	4	0	0	0	4			
0 T U 7	0	0	0	1	ן 1			
OTU8	U	38	51	44	133			
OTU9	0	0	1	0	1			
O T U 10	0	0	0	1	1			
O T U 11	0	1	2	1	4			
<u>O T U 12</u>	0	0	0	1	1			
O T U 13	41	0	0	1	42			
Total	65	42	60	51	218			

CAP Related

Searching for other PAOs

FISH demonstration that new organism accumulates polyphosphate





FISH with "Dechloromonas" specific probe

DAPI showing polyphosphate accumulation (yellow)

Achieving very low effluent phosphorus concentrations

Simultaneous with very low nitrogen concentrations

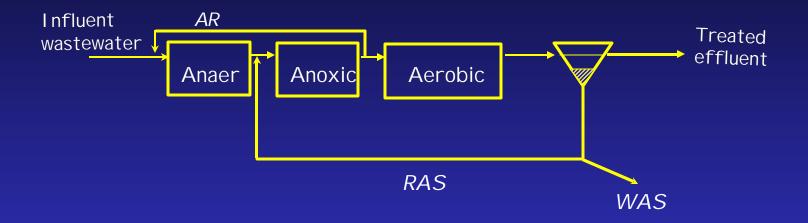
Nine Springs WWTP (Madison, WI)

- UCT process without nitrate recycle
- 42 MGD (158,000 m³/day)
- •Serves approx. 300,000 people

Converted to EBPR in 1996

Consistently achieves total phosphorus < 1 mg P/L in treated effluent

UCT w/o Nitrate Recycle



Concerns with EBPR at Nine Springs WWTP

Biosolids from EBPR have high P content

2.5% -> 5.0% (g P/kg VSS)
Too high for sustainable use in agriculture?





Pipe clogging and scale formation in anaerobic digesters

• Struvite

• Expensive maintenance

Concerns with EBPR at Nine Springs WWTP

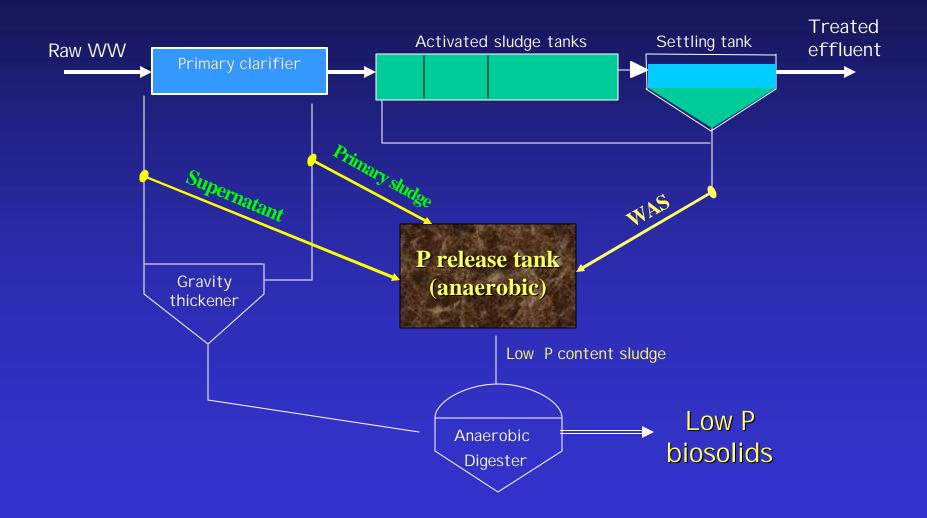
Suggested discharge limits: Total P = 0.05 mgP/L Total N = 3 mgN/L





Reduce phosphorus content in biosolids produced from EBPR reactors

• Evaluate process options to simultaneously achieve very low nitrogen and phosphorus effluents

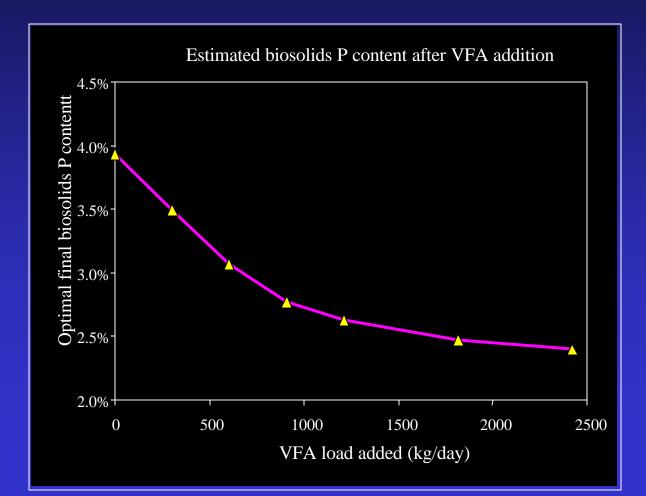


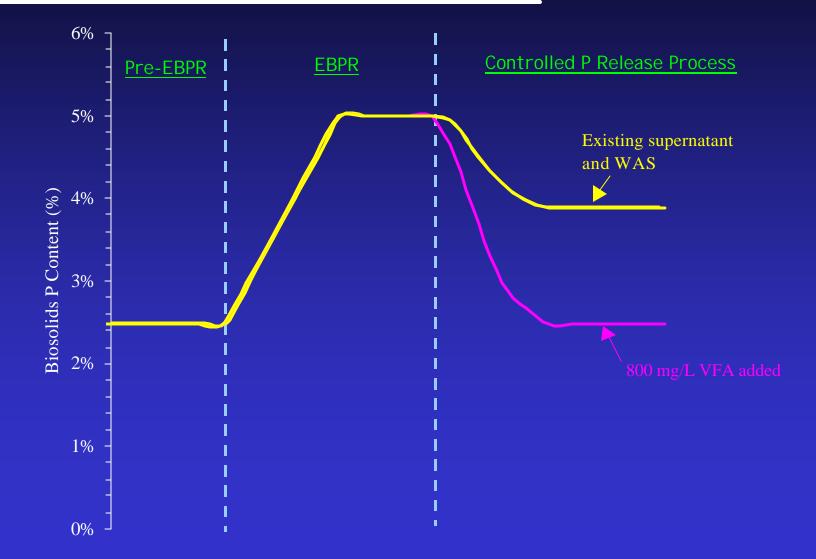
Phosphate release batch tests

WAS mixed with Primary sludge WAS mixed with Supernatant Phosphate release (mg P/L) 40 (mg P/L) 40 30 30 ^{>hosphate release} 20 20 10 10 $\left(\right)$ 0% 20% 40% 60% 80% 100% 0% 20% 40% 60% 80% 100% % WAS (by volume) % WAS (by volume)

Optimum P release at a 1:1 mixing ratio (by volume)

What if you have additional VFA available?





Achieving Low P and N

How low can we go?

Solids contribute to total P and total N

Is there enough biodegradable organic matter to support high levels of EBPR and denitrification?

 Contribution of residual organic matter to P and N

Achieving Low P and N

DEN

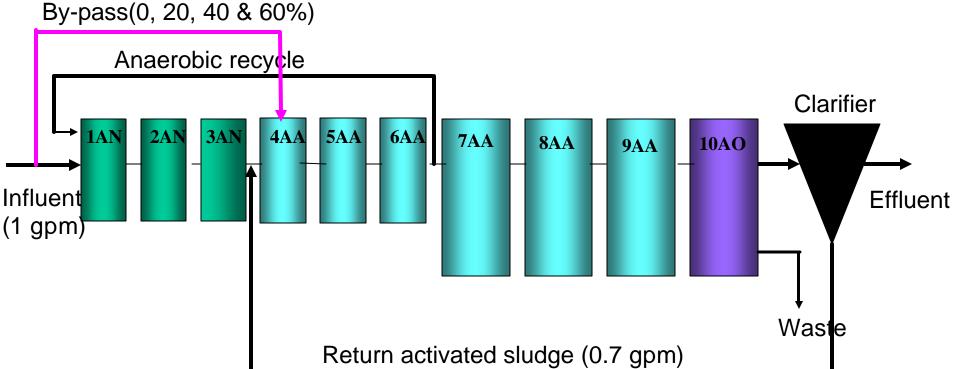
Pilot Plant simulating full-scale process (1 gpm)





Achieving Low P and N

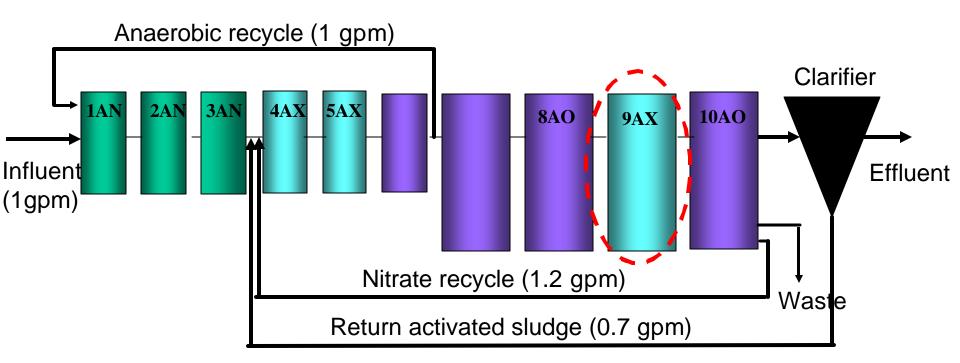
Pilot Plant configuration 1 (step feeding and aerated-anoxic conditions)



Volumes: AN = 80 gal, AA = 420 gal, AO = 130 gal

Achieving Low P and N

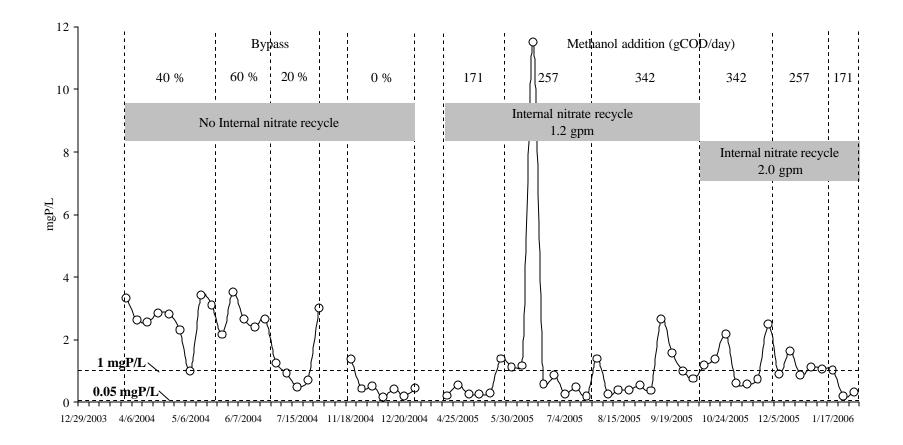
Pilot Plant configuration 2 (methanol, alum addition, and nitrate recycle)



Volumes: AN = 80 gal, AX1 = 60 gal, AO1 = 280 gal, AX2 = 130 gal, AO = 130 gal

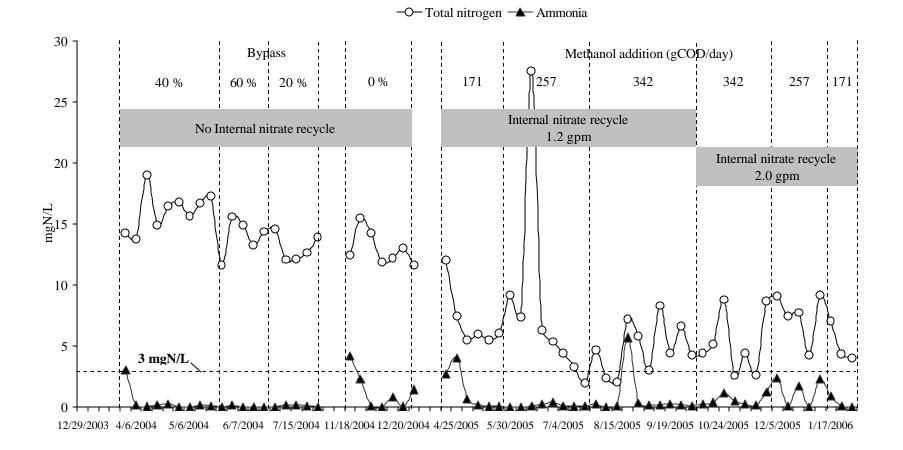
Step-feeding, nitrate recycle, and methanol

Phosphorus removal (unfiltered samples)



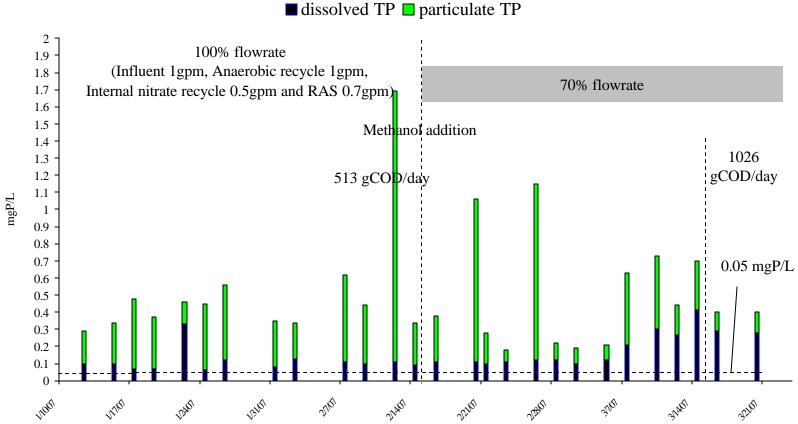
Step-feeding, nitrate recycle, and methanol

Nitrogen removal (unfiltered samples)



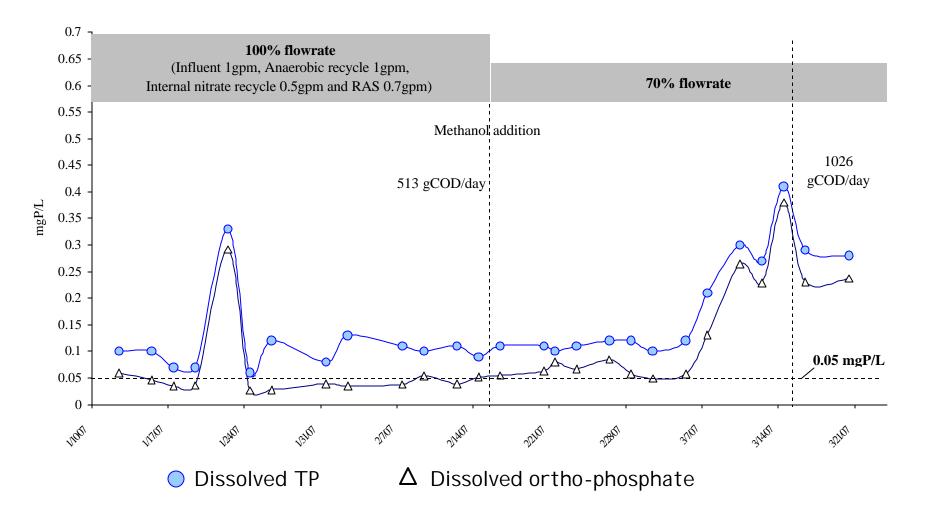
Methanol addition plus filtration

Effluent total phosphorus (filtered vs unfiltered)



Methanol addition plus filtration

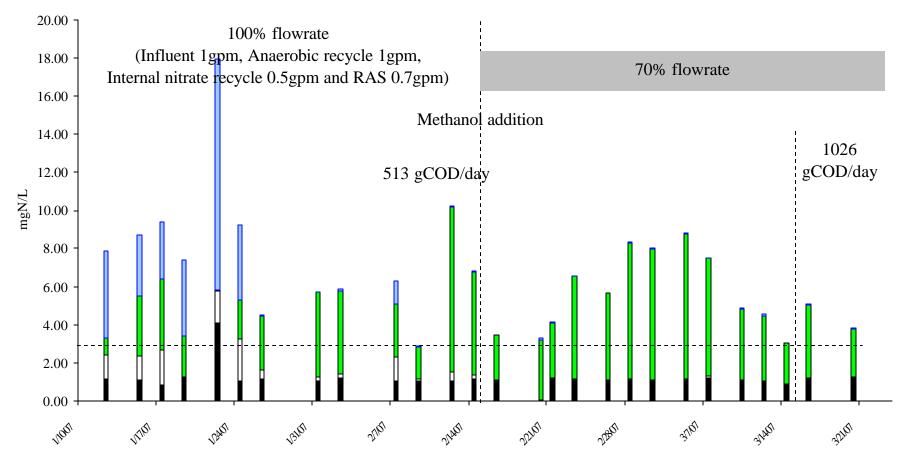
Effluent Dissolved total phosphorus (TP vs ortho-P)



Methanol addition plus filtration

Effluent Nitrogen



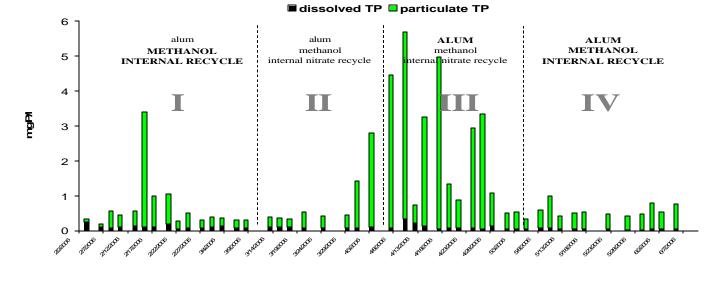


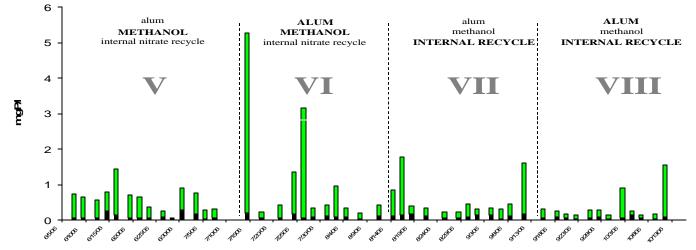
Alum addition

A factorial design (Methanol, alum, and nitrate recycle)

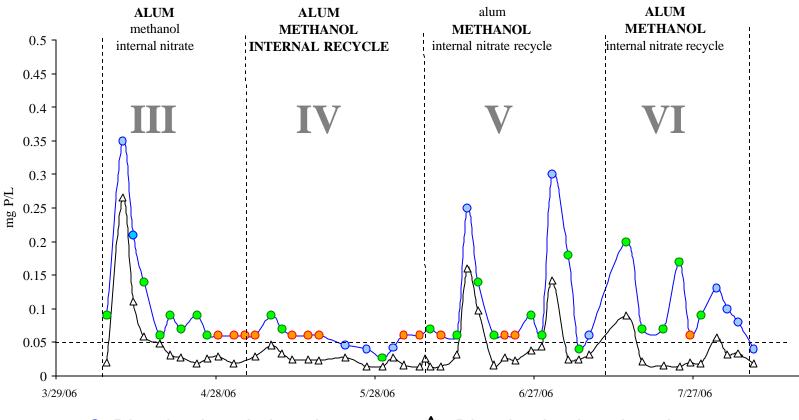
Factor	Low	High		Phase	Alum	Methanol	Nitrate recycle
		0		I	L	Н	Н
Alum	500	1000		II	L	L	L
(g/day)	500				Н	L	L
Methanol		510		IV	Н	Н	Н
(gCOD/day)	255			V	L	Н	L
		2.0		VI	Н	Н	L
Nitrate recycle (gal/min)	le 0.5			VII	L	L	Н
				VIII	Н	L	Н

Phosphorus (filtered vs non-filtered)



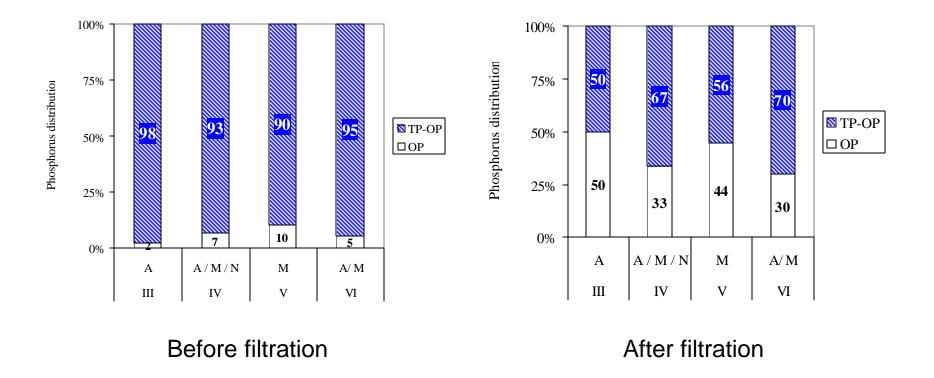


Effluent Total Phosphorus (filtered)



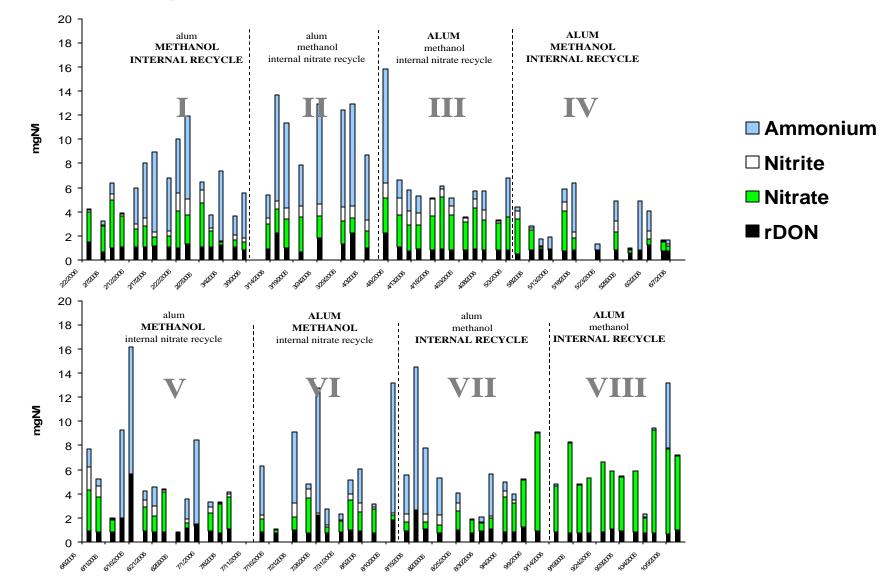
Dissolved total phosphorus
 Dissolved ortho-phosphate
 q: below limit of quantification (typically three times limit of detection)
 b: below limit of detection

Contribution of residual non-reactive phosphorus



(A: high level of alum; M: high level of methanol;N: high level of internal nitrate recycle)

Effluent Nitrogen (filtered)



Summary and Conclusions

CAP is an industrially relevant PAO

- CAP contribution to EBPR is not the same in all processes
- CAP in aerated-anoxic processes is not the main contributor

Evidence for a new PAO, related to Dechloromonas, in aerated-anaerobic EBPR

- EBPR implementation will increase P content of biosolids
- Biosolids P content can be controlled after EBPR

Summary and Conclusions

Limits of 0.05 mgP/L and 3 mgN/L are very difficult to achieve

Total Phosphorus (mg/L)

- Without filtration (EBPR only) ~ 0.7 (spikes 2.5)
- EBPR plus filtration
- EBPR, alum and filtration
- Residual ortho-phosphate
- Residual non-reactive phosphate

Total Nitrogen (mg/L)

With methanol addition ~ 5 mg/L

- ~ 0.2 (spikes 0.5)
- ~ 0.1 (spikes 0.4)
 - ~ 0.04
 - ~ 0.06

Acknowledgments

UW Madison

- Jason Hung
- Jordan Peccia
- Julie Zilles
- Patricia Sanhueza
- Sean Chaparro
- Ramesh Goel
- Daewook Kang
- Jason Flowers
- Hengky Harmita

<u>Madison Metropolitan</u> <u>Sewerage District (MMSD)</u>

<u>Funding</u>

- MMSD
- NSF
- WERF
- GERS (College of Engineering)